Interrupted Polymerization of Acrylates: Sequential Michael-Michael-Dieckmann Cyclizations for Easy, One-Pot, 2 + 2 + 2 Construction of Polyfunctionalized Cyclohexanones

Gary H. Posner* and Ellen M. Shulman-Roskes

Department of Chemistry, The Johns Hopkins University, Baltimore, Maryland 21218

Received February 13, 1989

Summary: In THF at -78 °C, lithio enolates of several different acyclic and cyclic carboxylate esters react cleanly with 2 equiv of an acrylate ester via a Michael-Michael-Dieckmann cyclization sequence to form polyfunctionalized cyclohexanones in 45-72% overall yields; this process represents useful, controlled interruptions of polymerization reactions.

Sir: Anionic polymerization of acrylic and methacrylic acid esters is a well-known industrial process of worldwide chemical interest.¹ Development of group-transfer polymerization (GTP) at Du Pont using silvl ketene acetals as initiators has led recently to considerable understanding of the mechanistic details of this commercially important, controlled, living polymerization.^{1a,2} Even more recently, metal-free (e.g. tetraalkyl ammonium) anionic polymerization of α -activated olefins has been reported, like GTP, to form valuable macromolecules of defined molecular weight and of narrow molecular weight distribution.^{1g} Like the back-biting step which terminates anionic polymerization of methacrylates,^{1e,3} termination of the GTP process also involves a final cyclization step (eq 1).² Much effort



now is being devoted internationally to understanding those factors that influence the balance between anionic chain propagation (i.e. polymerization) vs termination (i.e. cyclization).^{1e} From our interest in developing synthetic methods based on multicomponent annulations,⁴ we have found that polymerization of acrylates using lithium enolates of acetate esters as initiators is effectively inter-

110, 4754. (e) This interrupted polymerization of acrylates was presented initially by G. H. Posner at the Italian Chemical Society Summer School of Gargnano, Garda Lake, Italy, June 16-21, 1986.



rupted after only 2 equiv of acrylate have reacted forming ultimately diverse cyclohexanone products 2 in good yields (eq 2). It seems likely that an oxophilic lithium cation. as illustrated by proposed structure 1 for the intermediate in this process, is crucial for promoting intramolecular Claisen ester condensation and therefore for stopping polymerization.^{1e,5} That the three-component ring-forming reaction shown in eq 2 proceeded smoothly is especially



^{(5) (}a) Fowells, W.; Schuerch, C.; Boey, F. A.; Hood, F. P. J. Am. Chem. Soc. 1967, 89, 1396. (b) Vancea, L.; Bywater, S. Macromolecules 1981, 14, 1321. (c) Kraft, R.; Müller, A. H. E.; Höcker, H.; Schulz, G. V. 1991, 14, 1321. (c) Krat, K.; Muller, A. H. E.; Hocker, H.; Schulz, C. V. Macromol. Chem. Rapid Commun. 1980, 1, 363. (d) Jeuck, H.; Müller, A. H. E. Ibid. 1982, 3, 121. (e) Müller, A. H. E. In Recent Advances in Anionic Polymerization; Hogen Esch, J., Smid, J. Eds.; Elsevier: New York, 1987; p 205. (f) Cf.: Lochmann, L.; Rodova, M.; Petranek, J.; Lim, D. J. 1971, 19 D. J. Polym. Sci., Polym. Chem. Ed. 1974, 12, 2295.

0022-3263/89/1954-3514\$01.50/0 © 1989 American Chemical Society

^{(1) (}a) Webster, O. W.; Hertler, W. R.; Sogah, D. Y.; Farnham, W. B.; RajanBabu, T. V. J. Am. Chem. Soc. 1983, 105, 5706. (b) Hertler, W. R.; Sogah, D. Y.; Webster, O. W.; Trost, B. M. Macromolecules 1984, 17, 1417. (c) Reetz, M. T.; Ostarek, R.; Prejko, K. E.; Arlt, D.; Bömer, B. Angew. Chem., Int. Ed. Engl. 1986, 25, 1108. (d) Webster, O. W.; Sogah, Angew. Chem., Int. Ed. Engl. 1966, 25, 1105. (d) Webster, U. w.; Sogain, D. Y. In Recent Advances in Mechanistic and Synthetic Aspects of Polymerization; Fantaville, M., Guyot, A., Eds.; Reidel: Dordrecht, 1987; p 3. (e) van Beylen, M.; Bywater, S.; Smets, G.; Szwarc, M.; Worsfold, D. J. Adv. Polym. Sci. 1988, 86, 87. (f) Reetz, M. T.; Ostarek, R. J. Chem. Soc., Chem. Commun. 1988, 213. (g) Reetz, M. Angew. Chem., Int. Ed. Engl. 1988, 27, 1373. (i) Hertler, W. R.; RajanBabu, T. V.; Overall, D. W.; Dedder, C. S. Scarb, D. W. L. Arc, Chem. 2005. (i) Later, M. R.; RajanBabu, T. V.; Overall, D. W.; Reddy, G. S.; Sogah, D. Y. J. Am. Chem. Soc. 1988, 110, 5841.
 (2) Brittain, W. J.; Dicker, I. B. Macromolecules, in press. We thank

Du Pont for a preprint of this paper.

⁽³⁾ Morton, M. Anionic Polymerization, Principles and Practice;

noteworthy because of the considerable potential for proton transfer from one ester α -CH unit to the various ester enolate intermediates in this process. This easy, one-pot, 2 + 2 + 2, Michael-Michael-Ring-Closure (MI-MIRC) process,⁴ involving a terminating Dieckmann cyclization⁶ with overall formation of three carbon-carbon bonds (2a-c), represents a simple, convenient, and potentially versatile synthetic method for joining three 2carbon components into 6-membered carbocycles of varied substitution pattern.40

The generality of this synthetic method is illustrated in Scheme I. For example, methyl phenylthioacetate reacted with lithium diisopropylamide (LDA) in THF at -78 °C and then with 2.2 equiv of methyl acrylate to form, after acidic workup, trisubstituted cyclohexanone 3 on a multigram scale as a mixture of stereoisomers in 72% vield (Scheme II). Lower yields of dimethyl ester cycloadduct 3 were obtained when the ester group of the initiator was different from the ester group of the acrylate; lithium alkoxides formed in the terminal Dieckmann cyclization caused transesterifications. Stereoisomeric β -keto esters 3 were decarboxylated (NaCl, DMSO, 150 °C, 4.5 h)⁷ to form stereochemically homogeneous, polyfunctional ketone 4 in overall 57% yield based on starting methyl phenylthioacetate. Alternatively, stereoisomeric α -thio ketones 3 were oxidized into the corresponding sulfoxides (MCPBA, $0 \rightarrow 25$ °C, 1 h);⁸ pyrolysis (refluxing benzene)⁹ produced a conjugated cyclohexenone having substantial dienol character. Dehydrogenation (DDQ, refluxing benzene)¹⁰ formed regiospecifically trisubstituted benzene 5 in overall 42% yield based on starting methyl phenylthioacetate.

(10) Turner, A. B. In Synthetic Reagents; Pizey, J. S., Ed.; Wiley: New York, 1977; Vol. 3, p 193.

Spirobicyclic compounds, often useful synthons¹¹ and characteristic of different families of biologically active natural products,¹² in general are not easily prepared directly from simple components. As shown in Scheme I, lithium enolates of several carbocyclic and heterocyclic carboxylate esters reacted cleanly with 2 equiv of an acrylate ester to form spirobicyclic adducts 6 in 45-64% yields. This one-pot, three-component coupling process represents interrupted polymerizations leading easily and directly to regiospecifically polyfunctionalized spirobicyclic adducts.

In summary, the results reported here demonstrate (1) that lithio acetates, in which the α -carbon atom is either unsubstituted or mono- or disubstituted, initiate polymerization of acrylate esters: (2) that in THF at -78 °C such polymerization is effectively interrupted by a terminating Dieckmann cyclization; and (3) that this three-component coupling process leads rapidly and conveniently on a multigram scale to regiospecifically tri- and tetrasubstituted 6-membered carbocycles including spirobicyclic adducts. We are continuing to explore the scope and applications of these interrupted polymerizations.

Acknowledgment. We thank the donors of Petroleum Research Fund, administered by the American Chemical Society, and the NSF (Grant CHE-86-07974) for financial support and T. Kishimoto and C. Oh of this department for some fine experimental help.

Supplementary Material Available: A complete experimental section including descriptive procedures and full characterization data (6 pages). Ordering information is given on any current masthead page.

Efficient Enantiospecific Synthesis of Key A-Ring Synthons for the Preparation of 1α ,25-Dihydroxyvitamin D₃ Using a Chromium(II)-Mediated Reaction

Susumi Hatakeyama, Hirotoshi Numata, Ken Osanai, and Seiichi Takano*

Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980, Japan Received March 6, 1989

Summary: Key A-ring synthons for the synthesis of 1α ,25-dihydroxyvitamin D₃ have been prepared efficiently from (R)-(-)-carvone by use of diastereoselective chromium(II)-mediated addition of an allylic halide to an aldehyde as a key step.

Sir: Of the known vitamin D_3 metabolites, 1α ,25-dihydroxyvitamin D_3 (1), is considered to be the most potent stimulator of calcitropic effects such as intestinal calcium absorption and bone calcium mobilization.¹ Recently, this

hormone has also been found to suppress proliferation and induce differentiation in human myeloid leukemia cells.² These potent biological activities therefore have stimulated efforts toward the synthesis of 1α , 25-dihydroxyvitamin D₃ $(1).^{2-4}$



^{(6) (}a) Schaefer, J. P.; Bloomfield, J. J. Org. React. 1967, 15, 1. (b) Tarnchompoo, B.; Thebtaranonth, C.; Thebtaranonth, Y. Synthesis 1986, 785. (c) Kodpinid, M.; Siwapinyoyos, T.; Thebtaranonth, Y. J. Am. Chem. Soc. 1984, 106, 4862. (d) Flavin, M. T.; Lu, M. C. Tetrahedron Lett. 1982, 2238. (c) Cianuaro M. A. Friedel P. Cianmarine, A.S. Lett. 1983, 2335. (e) Gianturco, M. A.; Friedel, P.; Giammarino, A. S. Tetrahedron 1964, 20, 1763.
(7) Krapcho, A. P.; Lovey, A. J. Tetrahedron Lett. 1973, 957.
(8) Block, E. Reactions of Organosulfur Compounds; Academic Press:

New York, 1978; p 16. (9) Trost, B. M. Acc. Chem. Res. 1978, 11, 453.

^{(11) (}a) Burke, S. D.; Martiashaw, C. W.; Saunders, J. O.; Dike, M. S. J. Am. Chem. Soc. 1982, 104, 872. (b) Corey, E. J.; Petrzilka, M.; Ueda, Y. Helv. Chim. Acta 1977, 60, 2294. (c) Eaton, P. E.; Joba, P. G.; Nyi, K. J. Am. Chem. Soc. 1980, 102, 6636.

^{(12) (}a) Marshall, J. A.; Brady, S. F. J. Org. Chem. 1970, 35, 4068-4077. (b) Anderson, N. H.; Galcone, M. S.; Syrdal, D. D. Terahedron Lett. 1970, 1759-1762. (c) Coxon, D. T.; Price, K. R.; Howard, B. Ibid. 1974, 2921-2924. (d) Hayakawa, Y.; Aoki, K.; Hireata, Y. Ibid. 1973, 4963-4966. (e) Marshall, J. A.; Brady, S. F.; Anderson, N. H. Prog. Chem. Org. Nat. Prod. 1974, 31, 283. (f) Cf.: Coates, R. M. ibid. 1976, 33, 73. (g) Posner, G. H.; Hamill, T. G. J. Org. Chem. 1988, 53, 6031.

^{(1) (}a) Norman, A. W. In Vitamin D, the Calcium Homeostatic Steroid Hormone; Academic Press: New York, 1979. (b) DeLuca, H. F.; Paaren, H. E.; Schnoes, H. K. Top. Curr. Chem. 1979, 83, 1. (c) DeLuca, H. F.; Schnoes, H. K. Annu. Rev. Biochem. 1983, 52, 411. (d) DeLuca, H. F.; Schnoes, H. K. Annu. Rep. Med. Chem. 1984, 19, 179.